

Fig. 2. Comparison between the temperature dependences of the ratio S_{oo}/S_{11} of the integrated intensities of the ${}^5D_o \rightarrow {}^7F_o$ emission to the ${}^5D_1 \rightarrow {}^7F_1$ one in the spectra recorded under pulsed selective excitation into the 5D_1 level (\circ) and continuous ultraviolet excitation (+).

ficantly to the optical properties of the Sm²⁺: BaClF system at temperatures up to 300 K [4-6]. The physical explanation for these discrepancies lies in the fact that the feeding of levels ${}^{5}D_{0}$ and ${}^{5}D_{1}$ during the excitation pulse depends strongly on the pumping conditions [1]. Obviously, the role of the ${}^{5}D_{0} \rightarrow {}^{5}D_{1}$ radiationless transition will be enhanced if the population of the ${}^{5}D_{0}$ state at the end of the pulse is significant, as is the case for the ⁵D_o pumping. On the other hand, the contribution of this transition will be reduced if the ⁵D_o population remains equal to zero during the excitation pulse, as is the case for the ${}^{5}D_{1}$ or ${}^{5}D_{2}$ pumping. Under ultraviolet excitation, the feeding of the ${}^{5}D_{0}$ state during the pulse is ensured by fast non radiative processes connecting the 4f⁵-5d states directly to the ⁵D_o state at temperatures above 200 K as it clearly appears from the comparison between the temperature dependences of the ratio S_{00}/S_{11} of the integrated intensities of the ${}^{5}D_{o}/{}^{7}F_{o}$ emission to the ${}^{5}D_{1} \rightarrow {}^{7}F_{1}$ one in the spectra recorded under selective excitation into the ⁵D₁ level on one hand, and ultraviolet excitation on the other hand (Fig. 2).

Thus, the thermalization near room temperature between two rare-earth levels separated by an energy gap of several times the maximum phonon frequency in the crystal (294 cm⁻¹ in BaClF) [6] may be observed or not, depending on the optical pumping frequency. Similar investigations in other rareearth systems such as Eu^{3+} doped materials, with larger ${}^{5}D_{1} - {}^{5}D_{0}$ gaps and higher $4f^{5}$ -5d state energies should be of interest to complement this observation.

- 1 J. G. Gâcon, M. F. Joubert, M. Blanchard and B. Jacquier, *Phys. Rev., (to appear).*
- 2 A. S. M. Mahbub'ul Alam and Baldassare di Bartolo, Phys. Rev. Letters, 19, 1030 (1967).
- 3 A. S. M. Mahbub'ul Alam and Baldassare di Bartolo, J. Chem. Phys., 47, 3790 (1967).
- 4 J. C. Gâcon, J. C. Souillat, J. Sériot and B. di Bartolo, Phys. Status Solidi A39, 147 (1977).

- 5 F. Gaume, J. C. Gâcon, J. C. Souillat, J. Sériot and B. di Bartolo, in 'The Rare-Earths in Modern Science and Technology', Plenum, New York, 1978.
- 6 J. C. Gâcon, 'State Doctorate Thesis', Lyons University, unpublished (1978).

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Mechanisms for f-f Transition Probabilities in Lanthanide Coordination Compounds

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The Laporte-forbidden transitions of Ln(III) coordination compounds acquire a first-order electricdipole probability from transient dipoles induced in the ligand groups by an allowed even-multipole electric moment of the f-f excitation, and by the mixing of the f-f with f-d and f-g electron promotions under the electrostatic field of the ligands. Applied initially to Ln(III) complexes containing monoatomic ligands which have an effective isotropic polarizability the ligand-polarization mechanism is found to depend, on extension to the corresponding polyatomic ligand cases, upon the anisotropy of the ligand polarizability tensor in complexes belonging to the higher non-centric symmetries. The electrostatic field and the ligand polarization mechanisms make complementary intensity contributions to the f-f transitions of a given Ln(III) complex type, dependent upon the rank of the leading electric multipole moment. The polarization mechanism contributes principally to the intensities of the ligandhypersensitive 2^2 -pole f-f transitions, whereas the electrostatic mechanism is predominant for the 2⁶-pole transition intensities, and makes the more important contribution in the 2⁴-pole cases.

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Absorption and Fluorescence Spectra of Europium-(III) Compounds in Non-Aqueous Solutions

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The absorption and fluorescence spectra of Eu(III) compounds were investigated at room temperature in different non-aqueous solutions.